

6.02×10^{23}
AVOGADRO'S NUMBER

HOW BIG IS A MOLE?



Venue: Multipurpose hall
Time: 9:30 am



Organized by
Department of Chemistry
Indian Institute of Space Science and Technology, Thiruvananthapuram

International Mole Day 2026

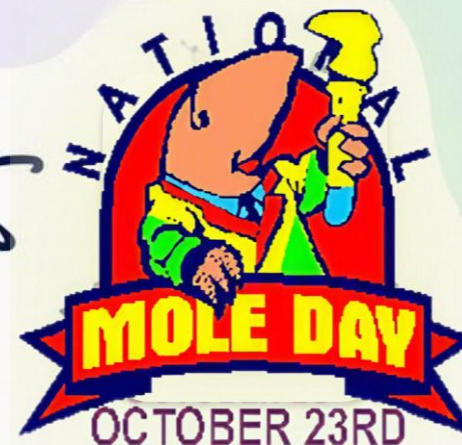
2026 THEME:
NMDF Molar Cup!

6.022×10^{23}

AVOGADRO CONSTANT
particles per mole



Just 18.0 g of water
contains about
 6.022×10^{23}
H₂O molecules!



Tiny
particles.
Huge numbers.
Infinite
possibilities.

